

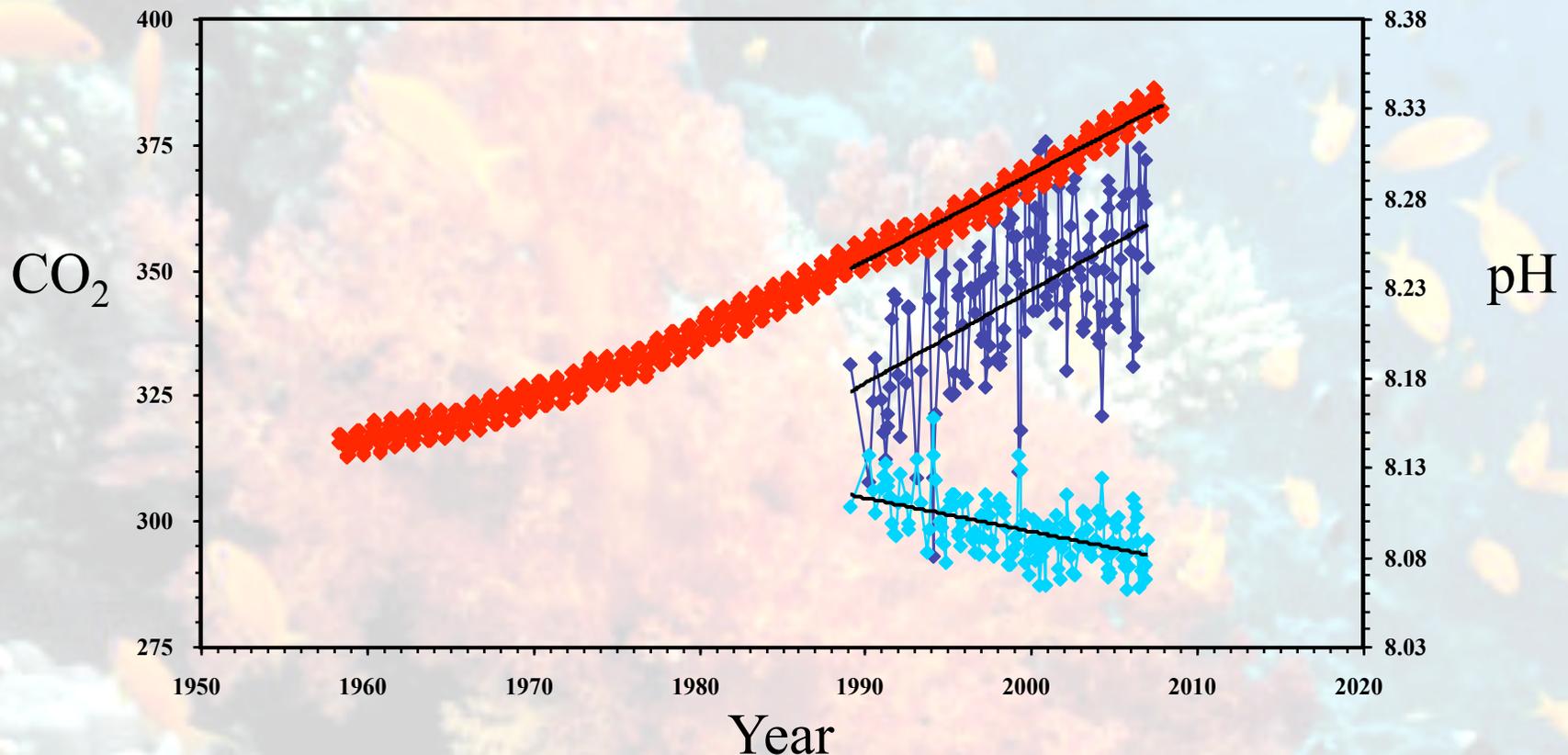
# Climate Change, Natural Resources and Coastal Management

## Ocean Acidification: The Other CO<sub>2</sub> Problem

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January 30, 2009

*With special thanks to: James Orr, Victoria Fabry, Carol Turley, Chris Sabine, Joanie Kleypas, Kitack Lee, and Simone Alin*



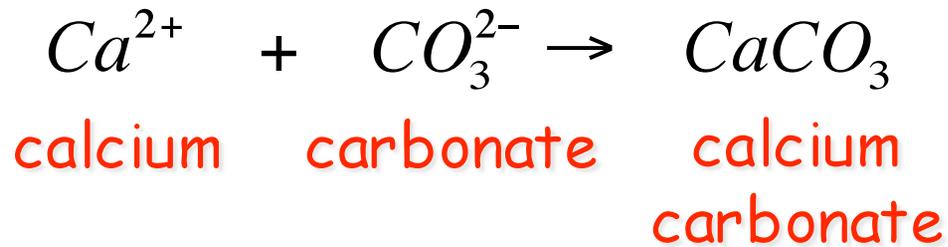


What we know about the ocean chemistry of *...saturation state*



*Saturation State*

$$\Omega_{phase} = \frac{[Ca^{2+}][CO_3^{2-}]}{K_{sp,phase}^*}$$



$\Omega > 1 = precipitation$

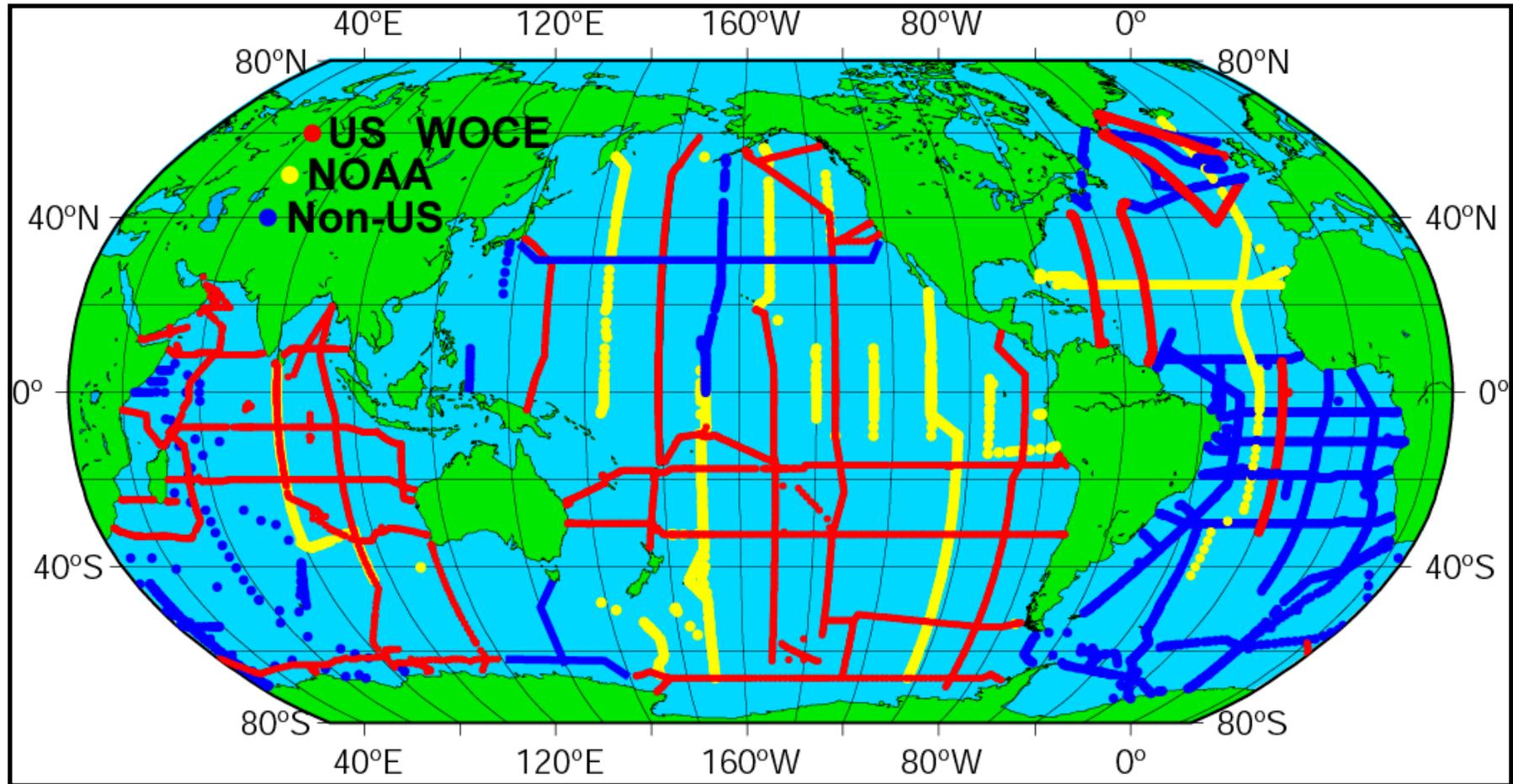
$\Omega = 1 = equilibrium$

$\Omega < 1 = dissolution$





## What we know about ocean $\text{CO}_2$ chemistry ...*from field observations*



WOCE/JGOFS/OACES Global  $\text{CO}_2$  Survey

~72,000 sample locations  
collected in the 1990s

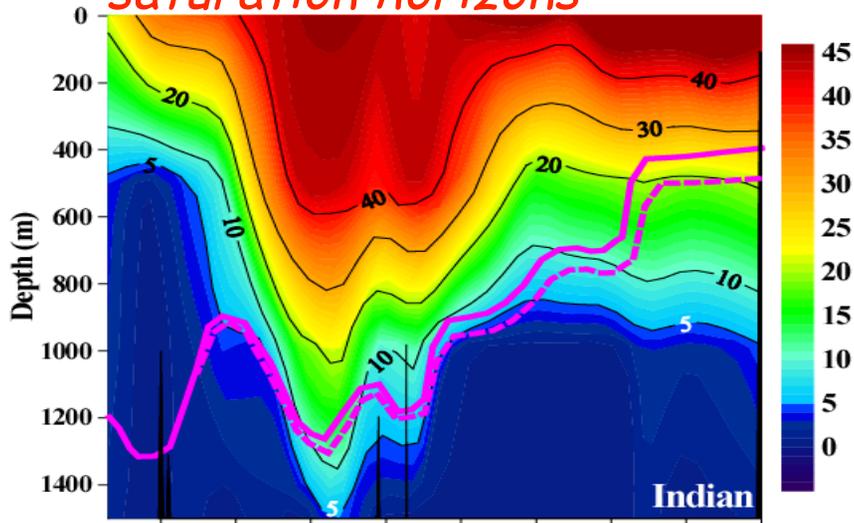
$\text{DIC} \pm 2 \mu\text{mol kg}^{-1}$

$\text{TA} \pm 4 \mu\text{mol kg}^{-1}$

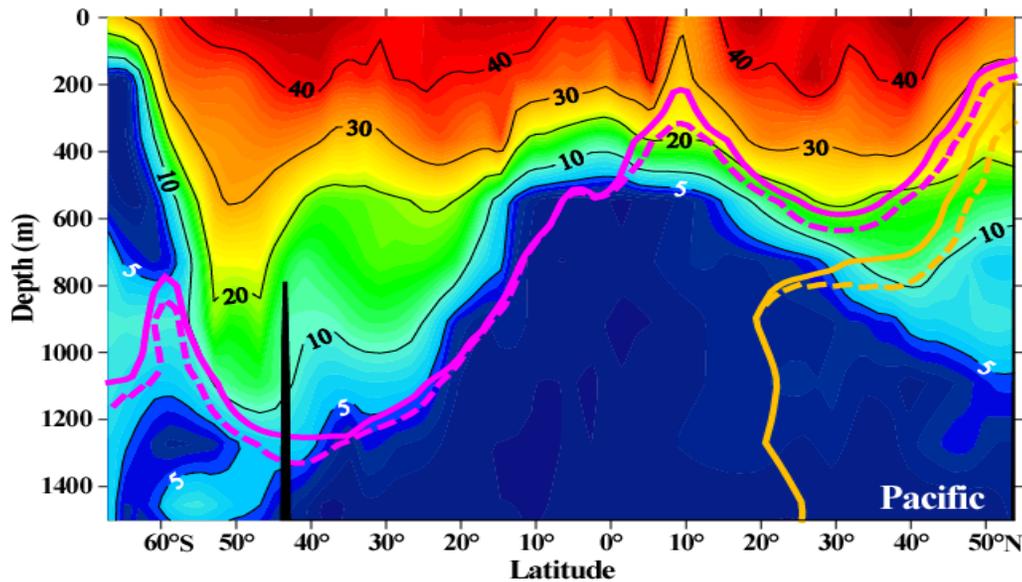
*Sabine et al (2004)*



# What we know about ocean CO<sub>2</sub> chemistry...*from observed shoaling saturation horizons*



Global Water-column  
Dissolution = 0.5 Pg C yr<sup>-1</sup>

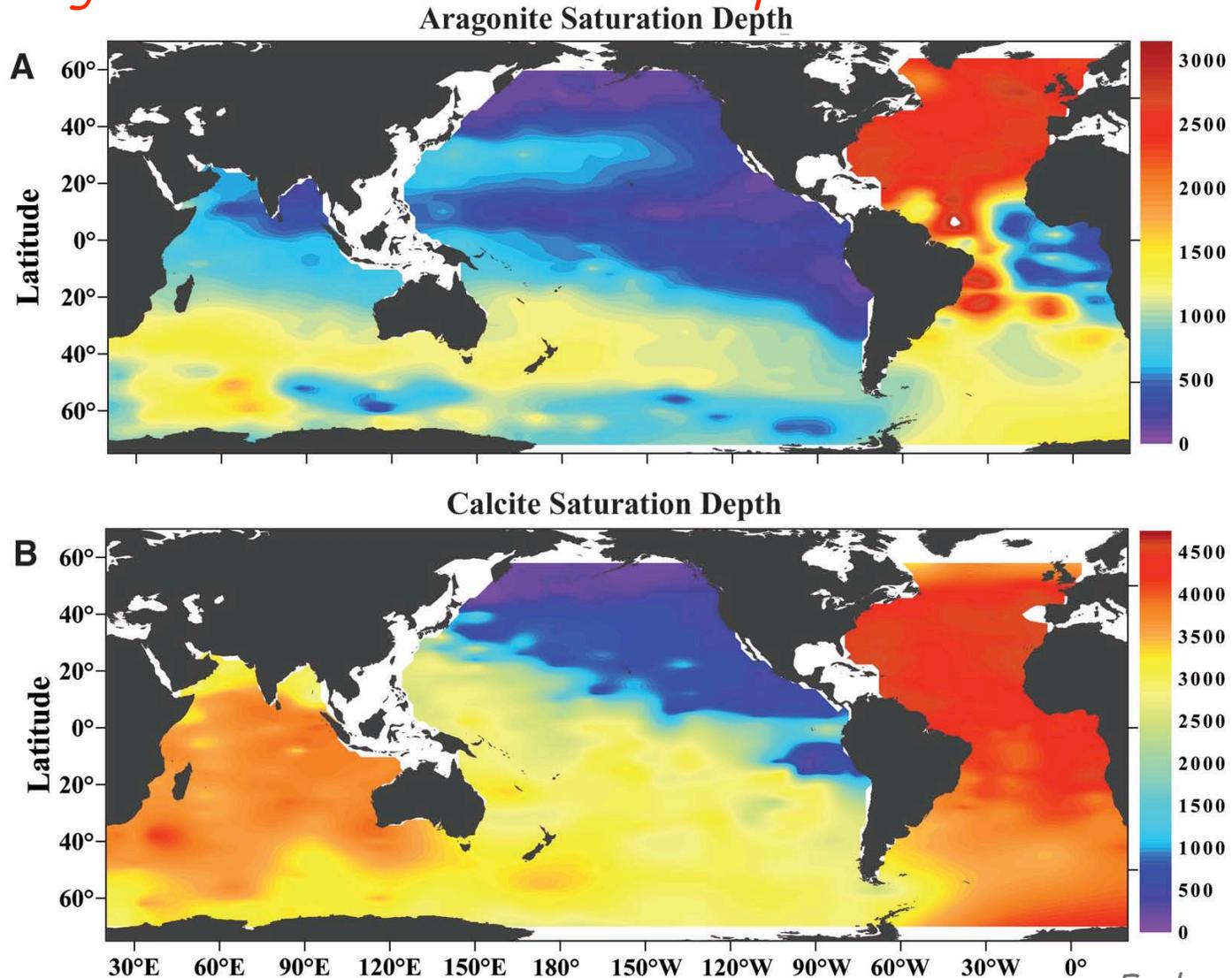


The aragonite and calcite saturation horizons have shoaled towards the surface of the oceans due to the penetration of anthropogenic CO<sub>2</sub> into the oceans.

Feely et al. (2004)



# What we know about ocean CO<sub>2</sub> chemistry...*from observed aragonite and calcite saturation depths*

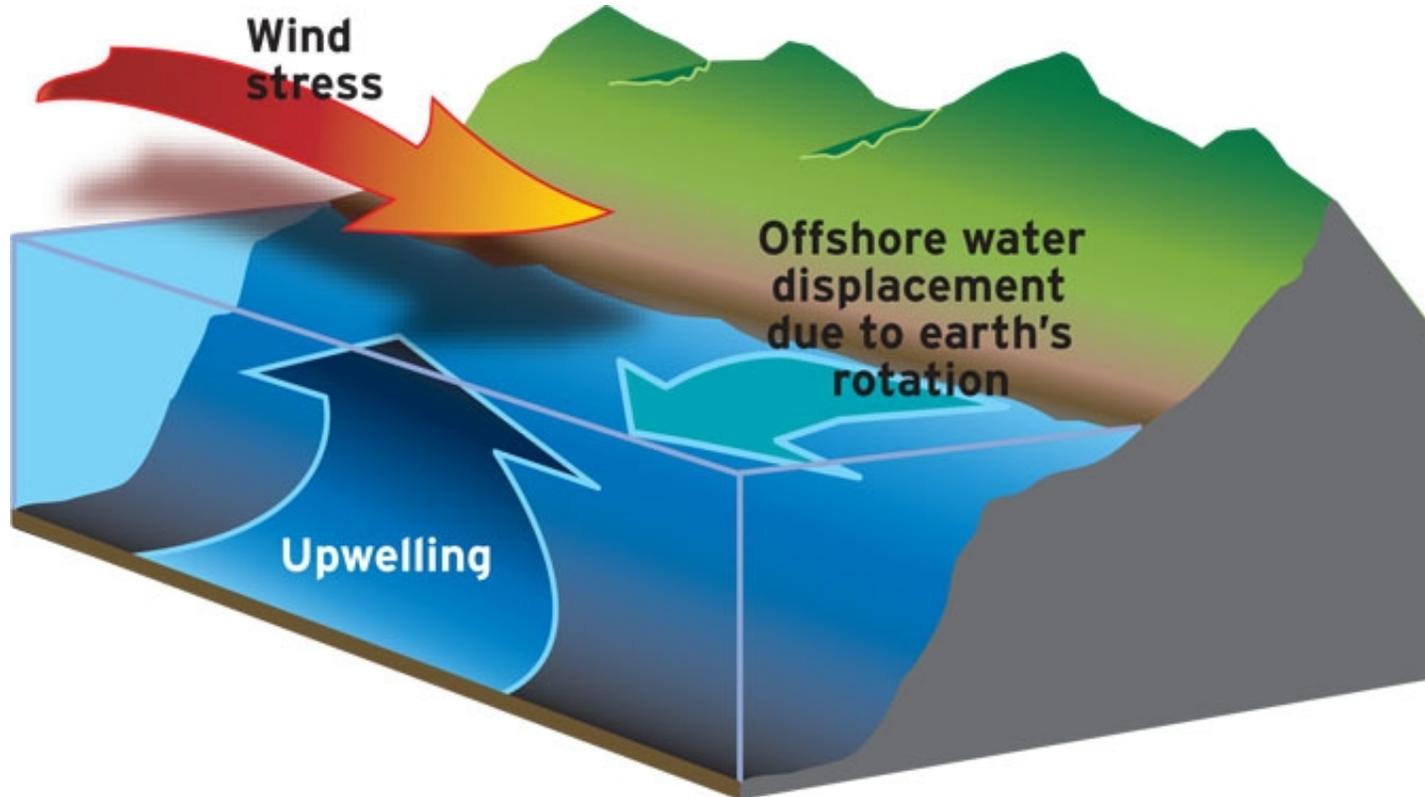


*Feely et al. (2004)*



## Natural processes that could accelerate the ocean acidification of coastal waters

### ➤ Coastal Upwelling





# North American Carbon Program

*Continental Carbon Budgets, Dynamics, Processes, and Management*



Newport



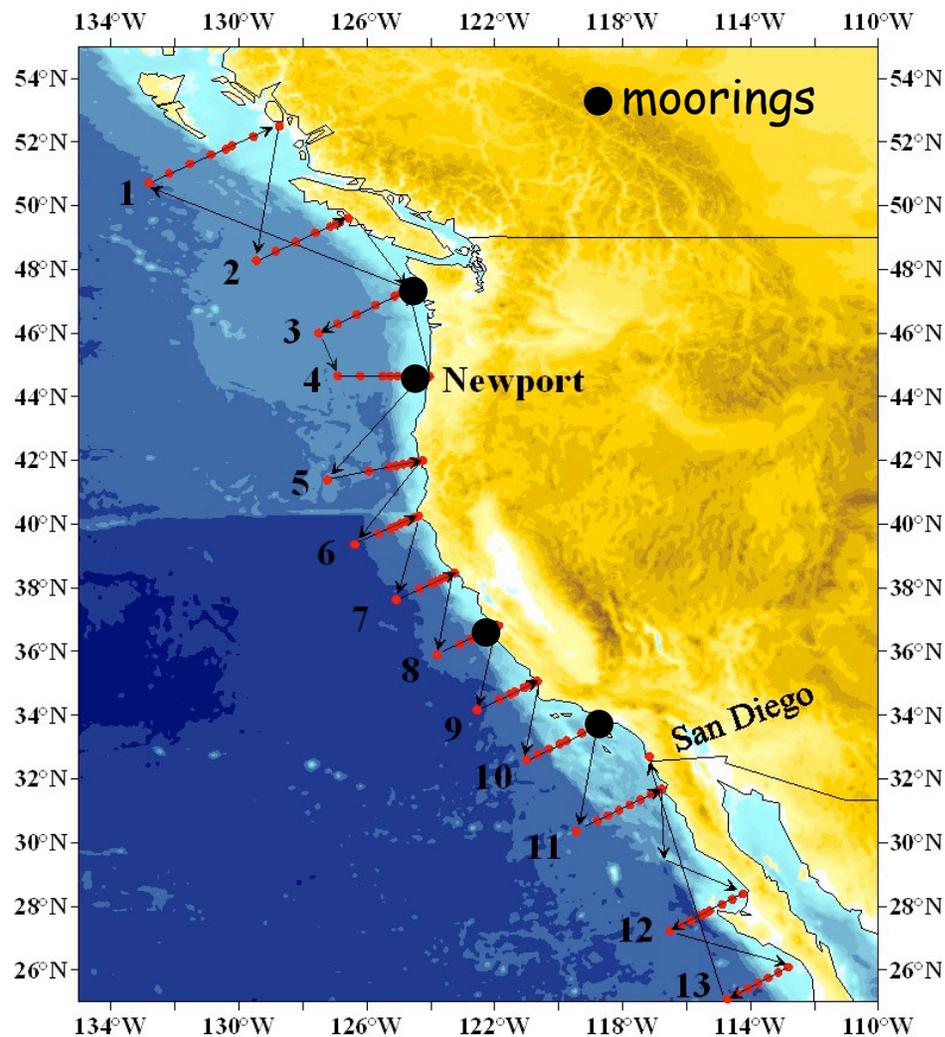
Aberdeen



MBARI



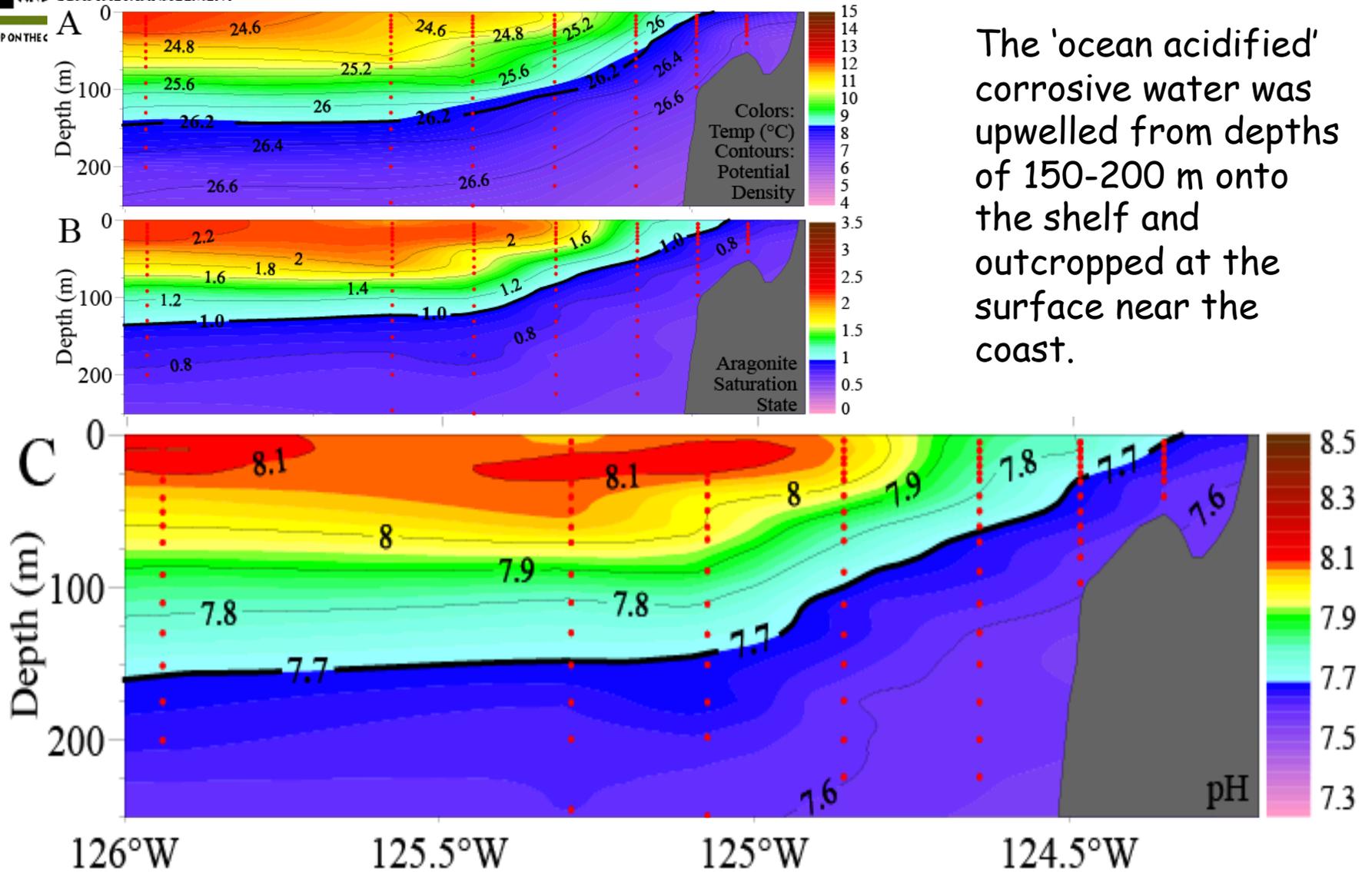
UCLA



NACP West Coast Survey Cruise : 11 May - 14 June 2007  
and mooring locations



A WORKSHOP ON THE C



The 'ocean acidified' corrosive water was upwelled from depths of 150-200 m onto the shelf and outcropped at the surface near the coast.

Red dots represent sample locations.

Vertical sections from Line 5 (Pt. St. George, California)

Feely et al. (2008)

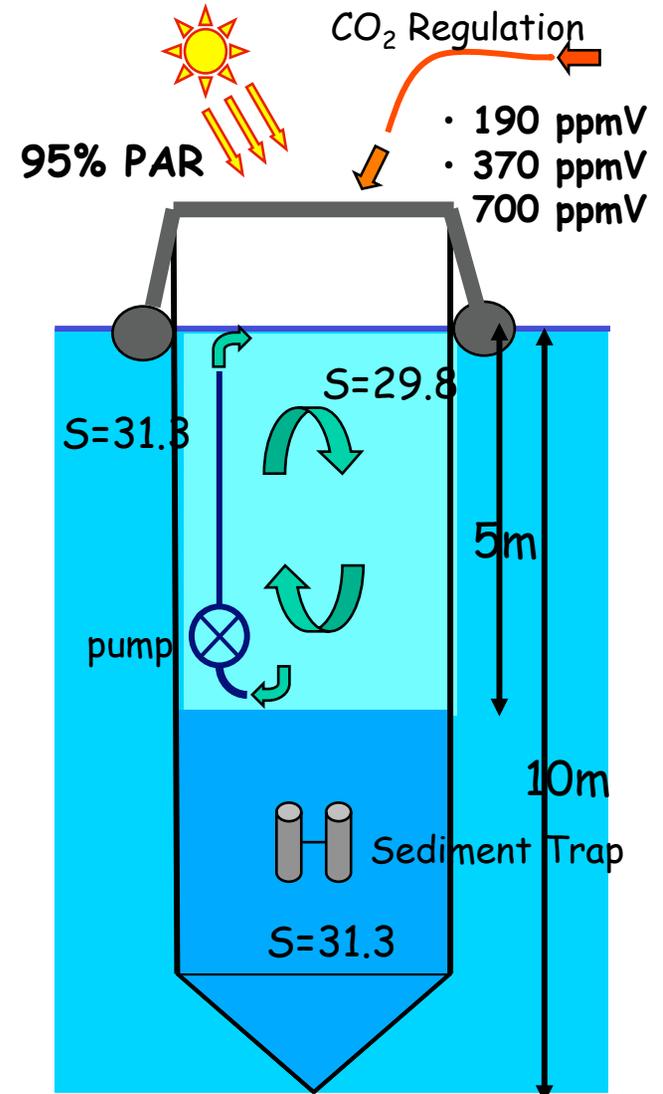


## Mesocosm experiment, Bergen Pelagic Ecosystem $\text{CO}_2$ Enrichment Study

Three  $\text{pCO}_2$  treatments representing:  
Glacial, Present, and Year 2100



Large Scale Mesocosm Facility, University of Bergen, Norway

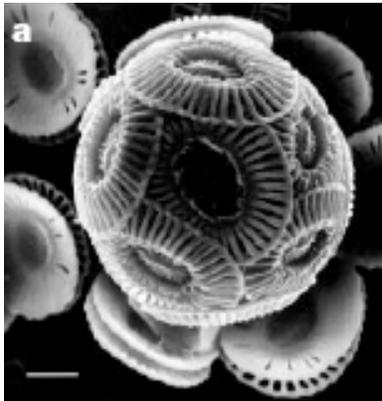


from U. Riebesell & B. Rost



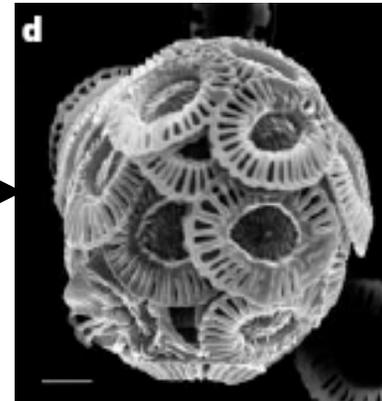
# Coccolithophore (single-celled algae)

$p\text{CO}_2$  280-380 ppmv



*Emiliana huxleyi*

$p\text{CO}_2$  780-850 ppmv

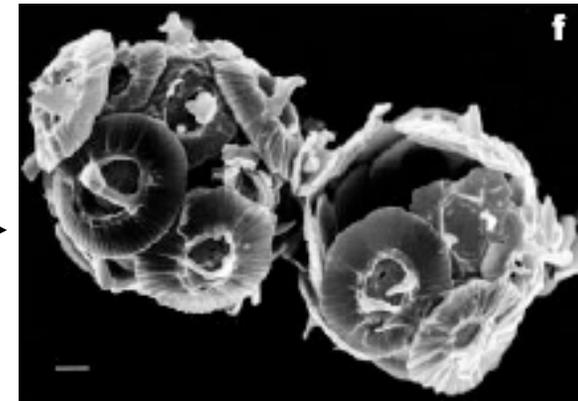


Calcification  
decreased

- 9 to 18%



*Gephyrocapsa oceanica*



- 45%

Malformed liths at high  $\text{CO}_2$

Manipulation of  $\text{CO}_2$  system by addition of HCl or NaOH



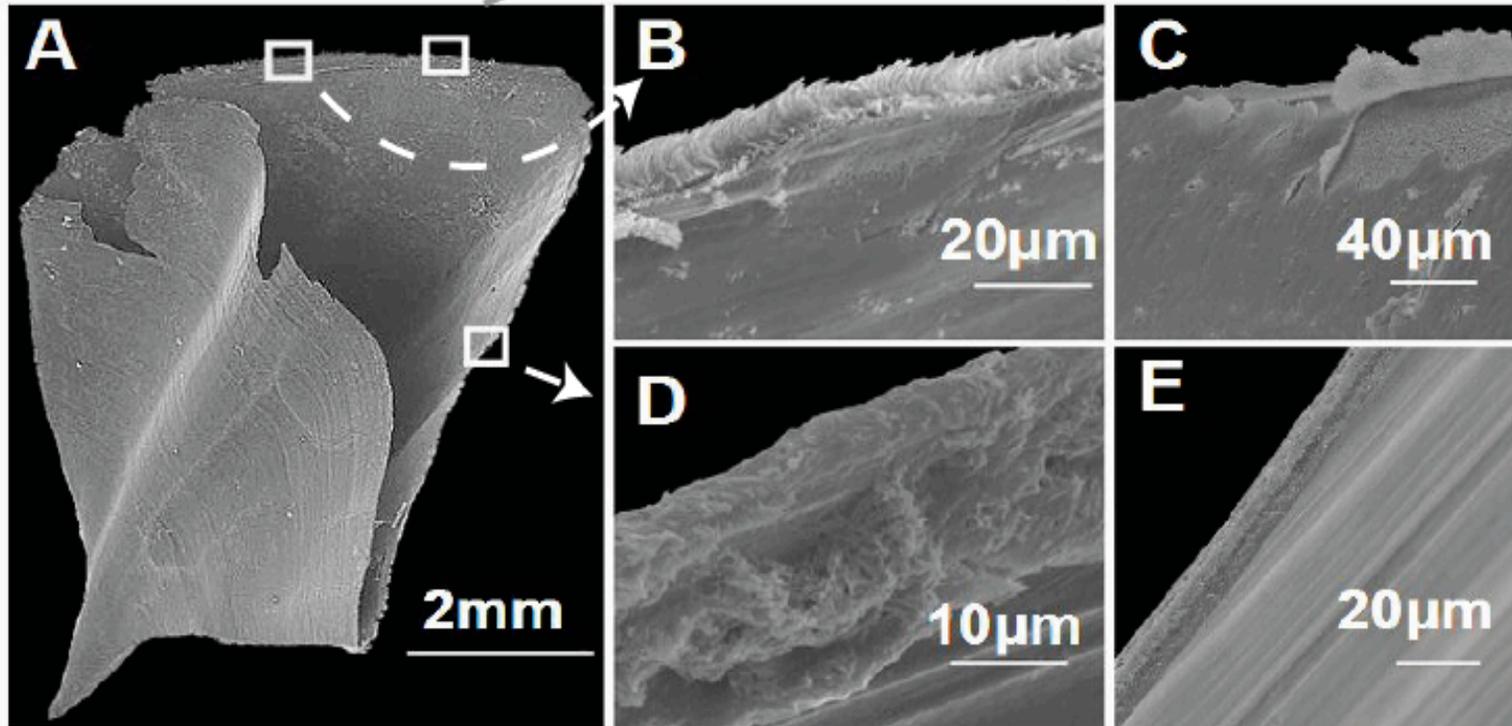
# Shelled Pteropods (planktonic snails)

Respiratory  $\text{CO}_2$  forced  $\Omega_A < 1$   
Shells of live animals start to dissolve within 48 hours

Whole shell:  
*Clio pyramidata*

Arag. rods exposed

Prismatic layer  
(1  $\mu\text{m}$ ) peels back



Aperture ( $\sim 7 \mu\text{m}$ ):  
advanced dissolution

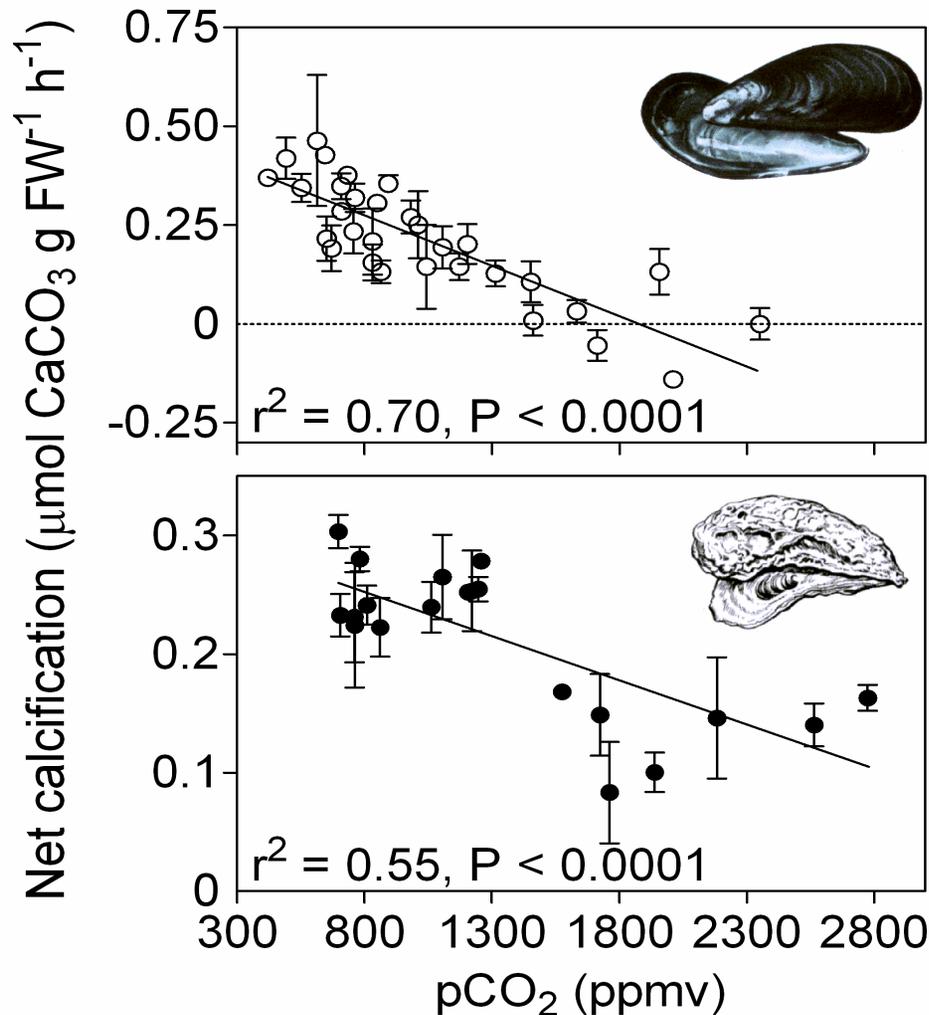
Normal shell: no  
dissolution

*Orr et al. (2005)*





## Response of mussels & oysters to elevated $\text{CO}_2$



Decrease in calcification rates for the 2 species:

*Mytilus edulis*

*Crassostrea gigas*

- Significant with  $\text{pCO}_2$  increase and  $[\text{CO}_3^{2-}]$  decrease

At  $\text{pCO}_2$  740 ppmv:

- 25% decrease in calcification for mussels

- 10% decrease in calcification for oysters



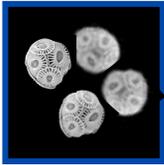
Ecologically and economically important organisms with planktonic larval stages

- **Bivalves:** clams, scallops, mussels, oysters
  - Valuable commercial fisheries
  - Mussels & oysters: ecosystem engineers
- **Echinoderms:** sea urchins, sea stars, sea cucumbers
  - Commercial fisheries: sea urchins & sea cukes
  - Sea stars: keystone species
- **Crustaceans:** shrimp, crabs, lobsters, copepods
  - Valuable commercial fisheries
  - Copepods: central role in marine food webs

Annual \$2  
Billion Dollar  
Industry



# Potential Effects on Open Ocean Food Webs



Coccolithophores



ARCOD@ims.uaf.edu

Copepods



Vicki Fabry

Pteropods



Barrie Kovish

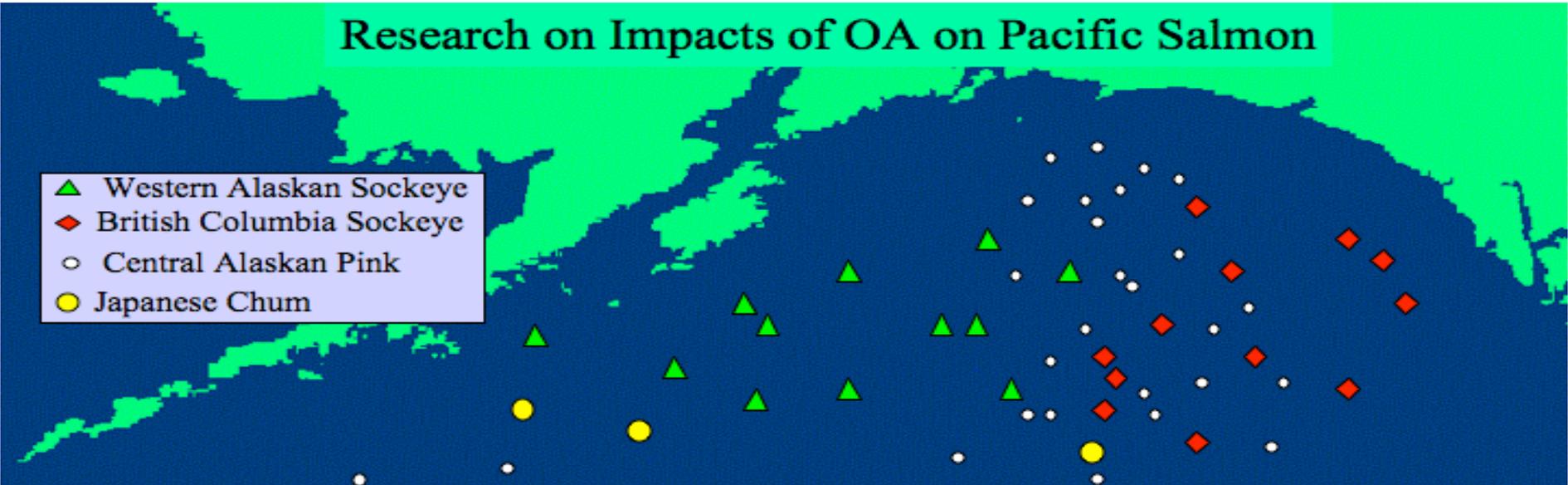
Pacific Salmon



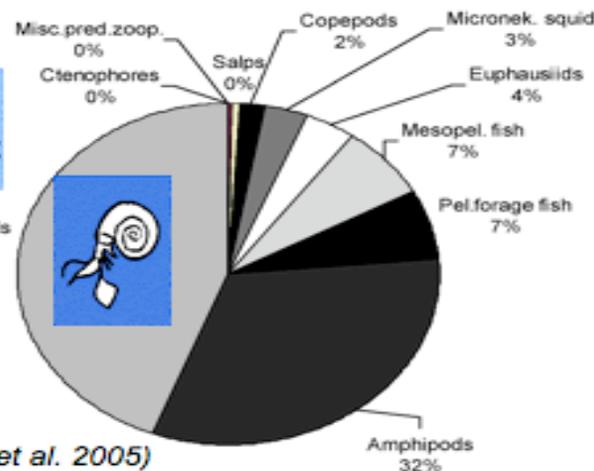
# What we know about the biological impacts of ocean acidification ...*on marine fish*

## Research on Impacts of OA on Pacific Salmon

- ▲ Western Alaskan Sockeye
- ◆ British Columbia Sockeye
- Central Alaskan Pink
- Japanese Chum



### Pink salmon diet



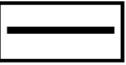
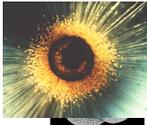
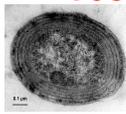
(Aydin et al. 2005)

### Predicted effect of climate change on pink salmon growth:

- 10% increase in water temperature leads to 3% drop in mature salmon body weight (physiological effect).
- 10% decrease in pteropod production leads to 20% drop in mature salmon body weight (prey limitation).



# Scorecard of Biological Impacts

Physiological process	Major group	# species studied	Response to increasing CO <sub>2</sub>			
						
<b>Calcification</b>						
	Coccolithophores	4	2	1	1	1
	Planktonic Foraminifera	2	2	-	-	-
	Molluscs	4	4	-	-	-
	Echinoderms	2	2	-	-	-
	Tropical Corals	11	11	-	-	-
	Coralline Red Algae	1	1	-	-	-
<b>Photosynthesis<sup>1</sup></b>						
	Coccolithophores <sup>2</sup>	2	-	2	2	-
	Prokaryotes	2	-	1	1	-
	Seagrasses	5	-	5	-	-
<b>Nitrogen Fixation</b>						
	Cyanobacteria	1	-	1	-	-
<b>Reproduction</b>						
	Molluscs	4	4	-	-	-
	Echinoderms	1	1	-	-	-

1) Strong interactive effects with nutrient and trace metals availability, light, and temperature  
2) Under nutrient replete conditions

Figure from Doney et al. (2009)



# Conclusions

- Impacts of ocean acidification on ecosystems are largely unknown.
- Calcification in many planktonic organisms is reduced at elevated  $CO_2$ , but the response is not uniform.
- Possible responses of ecosystems are speculative but could involve changes in species composition & abundances - could affect food webs, biogeochemical cycles.
- Baseline data with sufficient resolution are lacking in regions where  $CaCO_3$  saturation states are expected to decrease <1 over in next 50-100 years.